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SECTION	N 1 Chemicals and company ident	ification
	Product name	BaF2
	Product code	EB01
	Company name	CANON OPTRON INC.
	Address	1744-1, Kanakubo, Yuki-shi, Ibaraki-ken, 307-0015 Japan
	Section name	Sales Department
	Telephone number	+81-296-21-3700
	Fax number	+81-296-21-3770
	Emergency telephone tumber	+81-296-21-3700
	Use	Vacuum deposition material

SECTION 2 Hazards identification

GHS Classification (A classification by JIS Z 7252 "classification methods such as chemical substances based on GHS")

Physical hazards	Explosives	Classification not possible
	Flammable gases	Not applicable
	Aerosols	Not applicable
	Oxidizing gases	Not applicable
	Gas under pressure	Not applicable
	Flammable liquids	Not applicable
	Flammable solids	Classification not possible
	Self-reactive substances and mixtures	Classification not possible
	Pyrophoric liquids	Not applicable
	Pyrophoric solids	Classification not possible
	Self-heating substances and mixtures	Classification not possible
	Substances and mixtures which,in contact with water,emit flammable gases	Classification not possible
	Oxidizing liquids	Not applicable
	Oxidizing solids	Classification not possible
	Organic peroxides	Classification not possible
	Corrosive to metals	Classification not possible
	Desensitize explosives	Classification not possible
Health hazards	Acute toxicity(oral)	Category 3
	Acute toxicity(dermal)	Classification not possible
	Acute toxicity (Inhalation: Gases)	Not applicable
	Acute toxicity (Inhalation: Vapors)	Classification not possible

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	Acute toxicity (Inhalation: Dusts and mists)	Classification not possible
	Skin corrosion/irritation	Classification not possible
	Serious eye damage/eye irritation	Category 2A
	Respiratory sensitization	Classification not possible
	Skin sensitization	Classification not possible
	Germ cell mutagenicity	Classification not possible
	Carcinogenicity	Classification not possible
	Reproductive toxicity	Classification not possible
	Reproductive toxicity, effects on or via lactation	Classification not possible
	Specific target organ toxicity(single exposure)	Category 3
	Specific target organ toxicity(repeated exposure)	Category 1
	Aspiration hazard	Classification not possible
Environmental hazards	Hazardous to the aquatic environment Short-term(acute)	Classification not possible
	Hazardous to the aquatic environment Long-term(chronic)	Classification not possible
	Hazardous to the ozone layer	Classification not possible

Label elements

hazard Pictograms



Danger

Toxic if swallowed.

Causes serious eye irritation. May cause respiratory irritation.

Signal word

Dangerous goods hazard information

Precautionary statements

[Safety measures]

Do not breathe dust/fume/gas/mist/vapours/spray. Wash hands thoroughly after handling. Do not eat, drink or smoke when using this product. Wear Protective glovess/protective clothing/eye protection/face protection.

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【First−aid measures】	IF SWALLOWED: Immediately call poison center or doctor/physician. IF IN EYES: Rinse cautiously with water for several minutes.Remove contact lenses, if present and easy to do. Continue rising. IF exposed or concerned: Call poison center or doctor/physician. Get medical advice/attention if you feel unwell. Specific treatment . Rinse mouth. If eye irritation persists: Get medical advice/attention.
[Storage]	Store locked up.
【Disposal】	Dispose of contents/container in accordance with national regulations.
【Other hazards】	-

SECTION 3 Composition/information on ingredients

Substance/Mixture	Substance
Chemical name	Barium fluoride
Chemical formula	BaF2
Concentration or concentration range	99.9%<
CAS No.	7787-32-8
TSCA Inventry	Barium fluoride (BaF2)
EINECS number	232-108-0
Radioactive information	Radioactive substances are not used as the material. Therefore, there is no reason that ionizing radiation would be generated.
SECTION 4 First aid measures	
Inhalation	Remove person to fresh air and keep comfortable for breathing. Get medical advice/attention if you feel unwell.
Skin contact	Take off immediately all contaminated clothing. Rinse affected areas with water/shower. IF ON SKIN:Wash with plenty of soap and water. If skin irritation or rash occurs: :Get medical advice/attention.
Eye contact	Rinse cautiously with water for several minutes.Remove contact lenses, if present and easy to do. Continue rising. If eye irritation persists:Get medical advice/attention.
Ingestion	Rinse mouth. Get medical advice/attention.
Most important symptoms and effects, both acute and delayed	No data available
Protection of first aiders	Rescuers, wear suitable protective equipment as the situation demands.
Special precautions for physicians	No data available

SECTION	5 Firefighting measures	
S	Suitable extinguishing media	It use a water mist, dry chemicals, fire foam, carbon dioxide depending on the neighboring situation and the situation of the fire.
ι	Insuitable extinguishing media	Because a fire might spread through the outskirts, It avoid direct stick irrigation.
S	Specific hazards	In the case of fires, a toxic decomposition product may occur.
s	Specific extinguishing methods	It perform the fire fighting from windward. Restrict access to the area around the fire location to persons other than those involved with the fire. It move a container from the fire area if not dangerous.
	Special protective equipment for irefighters	On the occasion of fire extinguishing work, It wear appropriate personal protective equipment and rescue suit.
SECTION	6 Accidental release measures	
e	Personal precautions, protective equipment, and emergency procedures	It prohibit the entrance except the person concerned. The worker wears appropriate personal protective equipment (in item of "8.revelation prevention and protection measures" reference) and avoids eyes, contact and inhalation to skin.
E	nvironmental precautions	It avoid an outflow to the environmental average of the product to have possibilities to influence neighboring environment.
	I ethods and material for containment and cleaning up	It collects it in sky containers as if sweeping the scattered thing, and gathering you, or being able to absorb it with a vacuum sweeper, and from scattering not pitching a camp. The prohibition of handling and eating and drinking in neighboring of the storage area. It prevents the inflow to a drainage, a sewer, a basement or the closedown place.
	Secondary disaster prevention neasures	No data available
SECTION	7 Handling and storage	
F	Precautions for safe handling	
	Technical measures	Take measures for equipment as described in "8. Exposure controls/personal protection" and wear protective equipment.
	Safety handling precautions	It prevents you from producing dust.
	Avoidance of contact	Refer to "10. Stability and reactivity."
	Hygiene measures	Wash hands thoroughly after handling. Do not eat, drink or smoke when using this product.
	Conditions for safe storage, ncluding any incompatibilities	
	Safe storage conditions	It avoids direct rays of the sun and keeps it in the cool and dark space. Store in a well-ventilated place. Keep container tightly closed. Store locked up.

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Solubility

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Safety packaging material It use the container

It use the container which It can seal up without damage and the leak.

SECTION 8 Exposure controls/personal protection <u>BaF2</u> Permissible concentration Unestablished ACGIH Appropriate engineering controls In the work shop which dust produces. It use a device, an apparatus sealed up by all means or a local ventilator. Individual protection measures, such as personal protective equipment Respiratory protection Dustproof mask Hand protection Protective gloves Eye/face protection Dust-proof glasses Skin protection Protective clothing **SECTION 9** Physical and chemical properties Appearance Physical state Solid Form Pellets, granules Colour Transparent or white Odour None BaF2 1,353°C (Merck (15th, 2013)) Melting point/freezing point 2,260°C (Merck (15th, 2013)) Boiling point or initial boiling point and boiling range No data available Flammability Upper/lower flammability or No data available explosive limits Flash point No data available Noninflammability (GESTIS (2016)) Auto-ignition temperature Decomposition temperature No data available pН No data available Kinematic viscosity No data available

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1.586 g/L (10°C); 1.607 g/L (20°C); 1.620 g/L (30°C) (Merck (15th, 2013))	
No data available	
No data available	
No data available	
4.83 (Merck (15th, 2013))	
No data available	
No data available	
No data available	
	(Merck (15th, 2013)) No data available No data available 4.83 (Merck (15th, 2013)) No data available No data available

SECTION 10 Stability and reactivity

<u>BaF2</u>

Reactivity	It is stable under the normal handling condition.
Chemical stability	It is stable under the normal handling condition.
Possibility of hazardous reactions	A dangerous adverse effect is not caused under the normal handling condition.
Conditions to avoid	It avoid direct rays of the sun and keep it in the cool and dark space.
Incompatible materials	Oxidizer, reducing agent
Hazardous decomposition products	In the case of fires, a toxic decomposition product may occur.

SECTION 11 Toxicological information

<u>BaF2</u>

Acute toxicity(oral)	LD50 value of 250mg/kg oral study in rats (EHC 107, 1990)
Acute toxicity(dermal)	No data available
Acute toxicity (Inhalation: Gases)	Solid (GHS definition)
Acute toxicity (Inhalation: Vapours)	Solid (GHS definition)
Acute toxicity (Inhalation: Dusts and mists)	No data available
Skin corrosion/irritation	No data available
Serious eye damage/irritation	There are no data for this substance. However, based on a description that the fluorides cause irritation of the eyes (ACGIH (7th, 2001)), it was classified in Category 2.
Respiratory or skin sensitization	No data available
Germ cell mutagenicity	No data available

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Carcinogenicity	There is no carcinogenicity information on this substance or on the fluorides. However, as described in the same hazard class on the GHS classification of
	barium (CAS RN 7440-39-3), the EPA classified barium and its compounds as
	Group D or NL (IRIS (1998)) and ACGIH classified barium and its soluble
	compounds as A4 (ACGIH (7th, 2001)) based on results from animal tests using
	barium chloride dihydrate. Therefore, this substance was also classified as
	"Classification not possible" based on these classifications by other
	organizations.
Reproductive toxicity	Classification not possible due to lack of data. Besides, there is a description
	(EHC 107 (1990)) that in a test in which the substance was orally administered
	to pregnant rats on the first days of gestation, decreases in the survival of 5–
	day-old embryos and in newborn birth weights, and increase in mortality rate of
	newborn were observed, however, no incidence of malformation was observed.
Specific target organ toxicity(single In ACGIH (7th, 2001), there is a description that fluorides cause irritation to
exposure)	the respiratory tract. Therefore, this substance was classified in Category 3
	(respiratory tract irritation).
Specific target organ	There is no information on this substance itself.
toxicity(repeated exposure)	However, it is said that the toxicity of barium and barium compounds depends
	largely on the solubility, and that the toxicity is an inhibitory action of potassium channel by barium ions (ATSDR (2007)). The barium ions and the soluble
	compounds of barium (notably barium chloride, barium nitrate, barium hydroxide)
	are toxic to humans. Although barium carbonate is relatively insoluble in water, it
	is toxic to humans because it is soluble in the gastrointestinal tract. The
	insoluble compounds of barium (notably barium sulfate) are inefficient sources of
	the barium ion and are therefore generally nontoxic to humans (ATSDR (2007)).
	Therefore, this substance is thought to exhibit similar target organs as those exhibited by soluble barium.
	It is reported that incidences of hypertension, heart disease, and stroke
	increased in a population of neighborhood who ingested drinking water containing
	water soluble barium such as barium chloride, and that an increase in mortality due to heart diseases such as vascular disorder and arteriosclerosis was
	observed in other similar groups (ATSDR (2007)). For the inhalation route, a high
	rate of elevated blood pressure was reported among workers exposed to soluble barium (mean 1.07 g/ m^3) through work that involved blending and grinding severa
	grades of barium at a mineral processing facility (CICAD 33 (2001)). It is written
	that among the effects of soluble barium in humans, ingestion of high levels of
	soluble barium compounds may cause the following acute effects: gastroenteritis (vomiting, diarrhoea, abdominal pain), hypopotassemia, hypertension, cardiac
	arrhythmias, and skeletal muscle paralysis (CICAD 33 (2001)). Additionally, there are only several reported cases that renal failure and renal
	insufficiency occurred as acute barium poisoning. However, as for experimental
	animals, in 13-week or 2-year studies of barium dichloride or its dihydrate
	administered to rats or mice in drinking water, deaths attributed to nephropathy
	were observed at doses corresponding to Category 2 or higher (converted
	guidance value as barium dichloride: 271-803 mg/kg/day). It is described that
	the kidneys are the most sensitive target organ in the experimental animals (ATSDR (2007)). Therefore, the kidneys are thought to be one of the target
	organs in humans as well.
	Additionally, in the "Fluorides" of ACGIH, it is reported that bone lesions
	related to fluorosis caused by occupational exposure to inorganic fluorides (ACGIH (7th, 2001)).
	From the above, effects of soluble barium compounds including this substance
	were thought to have a high probability of occurring in the cardiovascular
	system, nervous system, muscular system, and kidneys. In addition, as a fluoride,
	this substance may have effects on the bones. Therefore, the substance was

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Aspiration hazard	No data available	
Other information	No data available	
SECTION 12 Ecological information		
	<u>BaF2</u>	
Toxicity		
Hazardous to the aquatic environment Short- term(acute)	No data available	
Hazardous to the aquatic environment Long- term(chronic)	No data available	
Persistence and degradablility	No data available	
Bioaccumulative potential	No data available	
Mobility in soil	No data available	
Hazard to the ozone layer	No data available	
Other adverse effects	No data available	
SECTION 13 Disposal considerations Waste treatment methods	Process is contracted to industrial waste disposers who received approval of a prefectural governor.	
Contaminated container and contaminated packaging	The container is recycled after being cleaned, or is appropriately processed according to the standards of related laws and regulations. When disposing of empty containers, the contents should be completely removed.	
SECTION 14 Transport information		
	BaF2	
International regulation		
UN number	1564	
UN proper shipping name	BARIUM COMPOUND, N.O.S.	
UN classification	6.1	
Transport hazard class	Not applicable	
Packing group	Ш	
Hazardous to the aquatic environment	No data available	
Maritime transport in bulk according to IMO instruments	No data available	
Japanese lows and regulations	Refer to "15. Regulatory information."	

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Special precautions for users	The Fire Services Act, Deleterious Substance Control Law, an object of the yellow card maintenance by the rule of the explosives control method. When transporting, protect from direct sunlight and take on cargo without breakage of container, corrosion and leakage. Do not stack heavy good thereupon.	
Special Provisions	154	

SECTION 15 Regulatoly information (Japan)

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Occupational Safety and Health Law	Not applicable
PRTR Law	Not applicable
Poisonous and Deleterious Substances control Law	There is it in the case of an application or an application
Labor Standards Act	There is it in the case of an application or an application
Chemical substances control Law	Not applicable
Fire fighting Law	There is it in the case of an application or an application
Air Pollution Control Act	Not applicable
Water Pollution Prevention Act	There is it in the case of an application or an application
Water Supply Act	There is it in the case of an application or an application
Sewerage Act	There is it in the case of an application or an application
Marine Pollution Prevention Law	Not applicable
Waste Management and Public Cleansing Act	Not applicable
Note	Ensure this material in compliance with federal requirements and ensure conformity to local regulations.

SECTION 16 Other information

The Safety Data Sheet (SDS) has been prepared based on currently available materials, information and data, and may be revised based on new information. Further, the important points in the SDS are made for the purpose of normal handling. When handling the user product in a specialized manner, take the appropriate safety measures for the application or method. Further, Canon Optron Inc. has paid sufficient attention to the described contents of the SDS, but does not guarantee the accuracy of its contents.

The SDS prepared by our company includes all findings from our investigation for reference. Not applicable to all items listed.

Literature Reference

[WEB site]

National Institute of Technology and Evaluation Homepage Japan Advanced Information Center of Safety and Health Homepage Ministry of Health, Labour and Welfare Homepage [Regulatory review Tools] ezCRIC (Japan Chemical Database Ltd)