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SECTION 1 Chemicals and company identification

> BaF2 Product name Product code **EB01**

CANON OPTRON INC. Company name

Address 1744-1, Kanakubo, Yuki-shi, Ibaraki-ken, 307-0015 Japan

Section name Sales Department Telephone number +81-296-21-3700 +81-296-21-3770 Fax number Emergency telephone tumber +81-296-21-3700

Use Vacuum deposition material

SECTION 2 Hazards identification

GHS Classification (A classification by JIS Z 7252 "classification methods such as chemical substances based on GHS")

Physical hazards **Explosives** Classification not possible

> Flammable gases Not applicable Aerosols Not applicable Oxidizing gases Not applicable Gas under pressure Not applicable Flammable liquids Not applicable

Flammable solids Classification not possible Self-reactive substances and mixtures Classification not possible

Pyrophoric liquids Not applicable

Pyrophoric solids Classification not possible Self-heating substances and mixtures Classification not possible Substances and mixtures which,in Classification not possible

contact with water, emit flammable

gases

Oxidizing liquids Not applicable

Oxidizing solids Classification not possible Organic peroxides Classification not possible Corrosive to metals Classification not possible Desensitize explosives Classification not possible

Health hazards Acute toxicity(oral) Category 3

> Acute toxicity(dermal) Classification not possible

Acute toxicity (Inhalation: Gases) Not applicable

Acute toxicity (Inhalation: Vapors) Classification not possible

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Acute toxicity (Inhalation: Dusts and

mists)

Classification not possible

Skin corrosion/irritation

Classification not possible

Serious eye damage/eye irritation

Category 2A

Respiratory sensitization

Classification not possible

Skin sensitization

Classification not possible

Germ cell mutagenicity

Classification not possible

Carcinogenicity

Classification not possible

Reproductive toxicity

Classification not possible

Reproductive toxicity, effects on or via

lactation

Classification not possible

Specific target organ toxicity(single

Category 3

exposure)

Specific target organ toxicity(repeated

Category 1

exposure)

Aspiration hazard

Environmental hazards

Hazardous to the aquatic environment

Classification not possible Classification not possible

Short-term(acute)

Hazardous to the aquatic environment

Classification not possible

Long-term(chronic)

Hazardous to the ozone layer

Classification not possible

Label elements

hazard Pictograms

Skull and crossbones

Health Hazard



Signal word

Danger

Dangerous goods hazard

information

Toxic if swallowed.

Causes serious eye irritation.

May cause respiratory irritation.

Precautionary statements

[Safety measures]

Do not breathe dust/fume/gas/mist/vapours/spray.

Wash hands thoroughly after handling.

Do not eat, drink or smoke when using this product.

Wear Protective glovess/protective clothing/eye protection/face protection.

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[First-aid measures] IF SWALLOWED: Immediately call poison center or doctor/physician.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact

lenses, if present and easy to do. Continue rising.

IF exposed or concerned: Call poison center or doctor/physician.

Get medical advice/attention if you feel unwell.

Specific treatment.

Rinse mouth.

If eye irritation persists: Get medical advice/attention.

[Storage] Store locked up.

[Disposal] Dispose of contents/container in accordance with national regulations.

[Other hazards] -

SECTION 3 Composition/information on ingredients

Substance/Mixture Substance

Chemical name Barium fluoride

Chemical formula BaF2

Concentration or concentration

range

99.9%<

CAS No. 7787-32-8

TSCA Inventry Barium fluoride (BaF2)

EINECS number 232-108-0

Radioactive information Radioactive substances are not used as the material. Therefore, there is no

reason that ionizing radiation would be generated.

SECTION 4 First aid measures

Inhalation Remove person to fresh air and keep comfortable for breathing.

Get medical advice/attention if you feel unwell.

Skin contact Take off immediately all contaminated clothing. Rinse affected areas with

water/shower.

IF ON SKIN: Wash with plenty of soap and water.

If skin irritation or rash occurs: :Get medical advice/attention.

Eye contact Rinse cautiously with water for several minutes.Remove contact lenses, if

present and easy to do. Continue rising.

If eye irritation persists: Get medical advice/attention.

Ingestion Rinse mouth.

Get medical advice/attention.

Most important symptoms and effects, both acute and delayed

No data available

Protection of first aiders Rescuers, wear suitable protective equipment as the situation demands.

Special precautions for physicians No data available

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SECTION 5 Firefighting measures

neighboring situation and the situation of the fire.

Unsuitable extinguishing media Because a fire might spread through the outskirts, It avoid direct stick irrigation.

Specific hazards In the case of fires, a toxic decomposition product may occur.

Specific extinguishing methods
It perform the fire fighting from windward.

Restrict access to the area around the fire location to persons other than those

involved with the fire.

It move a container from the fire area if not dangerous.

Special protective equipment for

firefighters

On the occasion of fire extinguishing work, It wear appropriate personal

protective equipment and rescue suit.

SECTION 6 Accidental release measures

Personal precautions, protective equipment, and emergency

procedures

It prohibit the entrance except the person concerned.

The worker wears appropriate personal protective equipment (in item of "8.revelation prevention and protection measures" reference) and avoids eyes,

contact and inhalation to skin.

possibilities to influence neighboring environment.

Methods and material for containment and cleaning up

It collects it in sky containers as if sweeping the scattered thing, and gathering you, or being able to absorb it with a vacuum sweeper, and from scattering not

pitching a camp.

The prohibition of handling and eating and drinking in neighboring of the storage

area.

It prevents the inflow to a drainage, a sewer, a basement or the closedown place.

Secondary disaster prevention

measures

No data available

SECTION 7 Handling and storage

Precautions for safe handling

Technical measures Take measures for equipment as described in "8. Exposure controls/personal

protection" and wear protective equipment.

Safety handling precautions It prevents you from producing dust.

Avoidance of contact Refer to "10. Stability and reactivity."

Hygiene measures Wash hands thoroughly after handling.

Do not eat, drink or smoke when using this product.

Conditions for safe storage, including any incompatibilities

Store in a well-ventilated place. Keep container tightly closed.

Store locked up.



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Safety packaging material

It use the container which It can seal up without damage and the leak.

SECTION 8 Exposure controls/personal protection

BaF2

Permissible concentration

ACGIH Unestablished

Appropriate engineering controls In the work shop which dust produces, It use a device, an apparatus sealed up by

all means or a local ventilator.

Individual protection measures, such as personal protective equipment

Respiratory protection

Dustproof mask

Hand protection

Protective gloves

Eye/face protection

Dust-proof glasses

Skin protection

Protective clothing

SECTION 9 Physical and chemical properties

Appearance

Physical state Solid

Form Pellets, granules

Colour Transparent or white

Odour None

BaF2

Melting point/freezing point

Boiling point or initial boiling point

and boiling range

1,353°C (Merck (15th, 2013))

2,260°C (Merck (15th, 2013))

Flammability

Flash point

Upper/lower flammability or

explosive limits

No data available No data available

No data available

Auto-ignition temperature

Decomposition temperature

Noninflammability (GESTIS (2016))

Decomposition temperatu

No data available

pН

No data available

Kinematic viscosity

No data available

Solubility

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1.586 g/L (10°C); 1.607 g/L (20°C); 1.620 g/L (30°C)

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(Merck (15th, 2013))

Other solvents No data available

Partition coefficient: n-

octanol/water

Water

No data available

Vapour pressure No data available

Density and/or relative density 4.83 (Merck (15th, 2013))

(Density)

Relative vapor density

Particle characteristics

Other information

No data available

No data available

No data available

SECTION 10 Stability and reactivity

BaF2

Reactivity It is stable under the normal handling condition.

Possibility of hazardous reactions A dangerous adverse effect is not caused under the normal handling condition.

Incompatible materials Oxidizer, reducing agent

Hazardous decomposition products In the case of fires, a toxic decomposition product may occur.

SECTION 11 Toxicological information

Vapours)

BaF2

Acute toxicity(oral) LD50 value of 250mg/kg oral study in rats (EHC 107, 1990)

Acute toxicity(dermal) No data available

Acute toxicity (Inhalation: Gases) Solid (GHS definition)

Acute toxicity (Inhalation: Solid (GHS definition)

Acute toxicity (Inhalation: Dusts and mists)

Skin corrosion/irritation No data available

Serious eye damage/irritation

There are no data for this substance. However, based on a description that the

fluorides cause irritation of the eyes (ACGIH (7th, 2001)), it was classified in Category 2.

Oatogory

Respiratory or skin sensitization No data available

Germ cell mutagenicity No data available

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Carcinogenicity

There is no carcinogenicity information on this substance or on the fluorides. However, as described in the same hazard class on the GHS classification of barium (CAS RN 7440-39-3), the EPA classified barium and its compounds as Group D or NL (IRIS (1998)) and ACGIH classified barium and its soluble compounds as A4 (ACGIH (7th, 2001)) based on results from animal tests using barium chloride dihydrate. Therefore, this substance was also classified as "Classification not possible" based on these classifications by other organizations.

Reproductive toxicity

Classification not possible due to lack of data. Besides, there is a description (EHC 107 (1990)) that in a test in which the substance was orally administered to pregnant rats on the first days of gestation, decreases in the survival of 5-day-old embryos and in newborn birth weights, and increase in mortality rate of newborn were observed, however, no incidence of malformation was observed.

Specific target organ toxicity(single exposure)

In ACGIH (7th, 2001), there is a description that fluorides cause irritation to the respiratory tract. Therefore, this substance was classified in Category 3 (respiratory tract irritation).

Specific target organ toxicity(repeated exposure)

There is no information on this substance itself.

However, it is said that the toxicity of barium and barium compounds depends largely on the solubility, and that the toxicity is an inhibitory action of potassium channel by barium ions (ATSDR (2007)). The barium ions and the soluble compounds of barium (notably barium chloride, barium nitrate, barium hydroxide) are toxic to humans. Although barium carbonate is relatively insoluble in water, it is toxic to humans because it is soluble in the gastrointestinal tract. The insoluble compounds of barium (notably barium sulfate) are inefficient sources of the barium ion and are therefore generally nontoxic to humans (ATSDR (2007)). Therefore, this substance is thought to exhibit similar target organs as those exhibited by soluble barium.

It is reported that incidences of hypertension, heart disease, and stroke increased in a population of neighborhood who ingested drinking water containing water soluble barium such as barium chloride, and that an increase in mortality due to heart diseases such as vascular disorder and arteriosclerosis was observed in other similar groups (ATSDR (2007)). For the inhalation route, a high rate of elevated blood pressure was reported among workers exposed to soluble barium (mean 1.07 g/m²) through work that involved blending and grinding several grades of barium at a mineral processing facility (CICAD 33 (2001)). It is written that among the effects of soluble barium in humans, ingestion of high levels of soluble barium compounds may cause the following acute effects: gastroenteritis (vomiting, diarrhoea, abdominal pain), hypopotassemia, hypertension, cardiac arrhythmias, and skeletal muscle paralysis (CICAD 33 (2001)).

Additionally, there are only several reported cases that renal failure and renal insufficiency occurred as acute barium poisoning. However, as for experimental animals, in 13-week or 2-year studies of barium dichloride or its dihydrate administered to rats or mice in drinking water, deaths attributed to nephropathy were observed at doses corresponding to Category 2 or higher (converted guidance value as barium dichloride: 271-803 mg/kg/day). It is described that the kidneys are the most sensitive target organ in the experimental animals (ATSDR (2007)). Therefore, the kidneys are thought to be one of the target organs in humans as well.

Additionally, in the "Fluorides" of ACGIH, it is reported that bone lesions related to fluorosis caused by occupational exposure to inorganic fluorides (ACGIH (7th, 2001)).

From the above, effects of soluble barium compounds including this substance were thought to have a high probability of occurring in the cardiovascular system, nervous system, muscular system, and kidneys. In addition, as a fluoride, this substance may have effects on the bones. Therefore, the substance was classified in Category 1 (cardiovascular system, nervous system, muscular system, kidney, bone).

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Aspiration hazard

No data available

SECTION 12 Ecological information

Other information

BaF2

Toxicity

Hazardous to the aquatic environment Shortterm(acute) Hazardous to the aquatic environment Longterm(chronic)

Persistence and degradablility

Bioaccumulative potential

Mobility in soil

Hazard to the ozone layer

Other adverse effects

No data available

SECTION 13 Disposal considerations

Waste treatment methods

Process is contracted to industrial waste disposers who received approval of a prefectural governor.

.....

Contaminated container and contaminated packaging

The container is recycled after being cleaned, or is appropriately processed according to the standards of related laws and regulations.

When disposing of empty containers, the contents should be completely removed.

SECTION 14 Transport information

BaF2

International regulation

UN number

1564

BARIUM COMPOUND, N.O.S.

UN classification

6.1

Transport hazard class Not applicable

Packing group

Ш

Hazardous to the aquatic environment

UN proper shipping name

No data available

Maritime transport in bulk according to IMO instruments

No data available

Japanese lows and regulations

Refer to "15. Regulatory information."

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Special precautions for users

The Fire Services Act, Deleterious Substance Control Law, an object of the yellow card maintenance by the rule of the explosives control method. When transporting, protect from direct sunlight and take on cargo without breakage of container, corrosion and leakage.

Do not stack heavy good thereupon.

Special Provisions 154

SECTION 15 Regulatoly information (Japan)

BaF2

Occupational Safety and Health Law	Not applicable
PRTR Law	Not applicable
Poisonous and Deleterious Substances control Law	There is it in the case of an application or an application
Labor Standards Act	There is it in the case of an application or an application
Chemical substances control Law	Not applicable
Fire fighting Law	There is it in the case of an application or an application
Air Pollution Control Act	Not applicable
Water Pollution Prevention Act	There is it in the case of an application or an application
Water Supply Act	There is it in the case of an application or an application
Sewerage Act	There is it in the case of an application or an application
Marine Pollution Prevention Law	Not applicable
Waste Management and Public Cleansing Act	Not applicable
Note	Ensure this material in compliance with federal requirements and ensure

conformity to local regulations.

SECTION 16 Other information

The Safety Data Sheet (SDS) has been prepared based on currently available materials, information and data, and may be revised based on new information. Further, the important points in the SDS are made for the purpose of normal handling. When handling the user product in a specialized manner, take the appropriate safety measures for the application or method. Further, Canon Optron Inc. has paid sufficient attention to the described contents of the SDS, but does not guarantee the accuracy of its contents.

The SDS prepared by our company includes all findings from our investigation for reference. Not applicable to all items listed.

Literature Reference

[WEB site]
National Institute of Technology and Evaluation Homepage
Japan Advanced Information Center of Safety and Health Homepage
Ministry of Health, Labour and Welfare Homepage
[Regulatory review Tools]
ezCRIC+ (Japan Chemical Database Ltd)

